

**REMARKS**

Reconsideration of this application is requested.

The pending claims are prior claims 1-12, as amended herein, and new claim 13.

Claim 1 has been amended to specify that the applicant's flavoring composition is a "cinnamon" flavor composition. This is consistent with the Examiner's suggested title and serves to underscore the novelty and unobviousness of the applicant's invention.

New claim 13 more specifically defines the applicant's flavor composition in "closed" fashion. Basis for the claim is found throughout the applicant's disclosure. See, for example, page 1, last ¶.

A new title has been provided as required.

The specification has also been amended to refer to the application's national phase status.

Additionally, the specification has been amended to include headings on the lines suggested by the Examiner.

Claim 11 has been amended to obviate the Examiner's objection to claims 11 and 12. Withdrawal of the objection is requested.

The claims have also been amended in view of the Examiner's Section 112, 2nd ¶ rejection of claims 3-6, 8 and 11-12. As amended, the claims are thought to be clear and definite and otherwise in acceptable form. Accordingly, the Examiner is requested to reconsider and withdraw the Section 112, 2nd ¶ rejection.

The Examiner is also requested to reconsider the Section 102(b) rejection of claims 1-4, 8-9 and 11-12 as anticipated by Bessette et al. (U.S. 6,004,569). The reference does not disclose the applicant's invention as defined by the rejected claims.

More specifically, the Examiner's attention is called to the fact that the applicant's claims specify a "flavour" composition, i.e. a composition suitable for use in flavoring, e.g. in oral care products. The Bessette et al. patent is concerned with pesticides. These clearly are not flavor compositions or suitable for use as flavoring materials. The applicant's compositions, therefore, differ fundamentally from the pesticides disclosed by Bessette et al.

The Examiner notes that Bessette et al. disclose pesticidal compositions comprising alpha-terpineol, eugenol and cinnamic alcohol in acetone. It is doubtful that one would use acetone in a flavor composition or for something to be used

orally. Furthermore, the Bessette et al. products include a significant amount of alpha-terpineol. Such products would not be suitable as flavor compositions.

The Examiner refers specifically to Sample 3B of Bessette et al. (Col. 5, lines 5-6) as showing a composition of alpha-terpineol, eugenol, cinnamic alcohol and acetone. However, Sample 3B does not anticipate the applicant's claims as the Sample 3B product contains too much alpha-terpineol to provide an acceptable flavor composition. Alpha-terpineol is widely used as a perfume ingredient due to its pine aroma and it can also be used in low levels as a flavor ingredient. Applicant attaches a copy of an extract from S. Arctander, Perfume and Flavor Chemicals (Montclair, N.J., 1969), which states near the end of the entry for alpha-terpineol that, when used as a flavor ingredient, alpha-terpineol is present at a level of from 5 to 40 ppm in the finished product. It is noted in this regard that the lowest level of alpha-terpineol disclosed in 3B of Bessette et al. is 15 wt% (when cinnamic alcohol is present at its maximum amount of 35 wt% and eugenol is present at its maximum amount of 50 wt%). From page 4, 3rd ¶ of the present application, the Examiner will see that the lowest level that the claimed flavor compositions will be present in a composition is 0.15 wt%, when the product is a mouthwash. Thus, if 3B were used at the lowest possible level in a mouthwash, it would result in a concentration of 0.0225 wt% (15% of 0.15%), or 225 ppm in the final product. This is more than 5 times greater than the maximum level typically used according to the Arctander reference. Such a product would, therefore, not be acceptable for consumption or oral use. The conclusion must, therefore, be that 3B is not a flavor composition because it contains too much alpha-terpineol. It is, accordingly, submitted, with respect, that applicant's claim 1 and consequently, claims 2-4, 8-9 and 11-12 define subject matter which is novel over, and not anticipated by, Bessette et al. Accordingly, the Examiner is requested to reconsider and withdraw the Section 102(b) rejection based on Bessette et al.

For basically similar reasons, the Examiner is requested to reconsider and withdraw the Section 102(b) rejection of claims 1-4 and 8-11 as anticipated by Slangan et al. (U.S. 3,917,870). The Examiner refers to Example VIII of Slangan to support the Section 102(b) rejection. This example describes a perfume formulation containing a host of components including cinnamic alcohol and eugenol together with a major amount of terpineol (400 parts by weight compared to 60 parts cinnamic alcohol and 20 parts eugenol). This large amount of terpineol means that the composition is not suitable as a flavor component for reasons noted above with respect to Bessette et al.

It is also noted that the materials heliotropin, aubepine and 4-(4-hydroxy-4-methylpentyl)-3-cyclohexene-1-carboxaldehyde (Lyrall), all mentioned as components in the perfume composition of Slangan's Example VIII, are only used in perfume formulations and never used in flavor compositions. Note in this regard the attached printout from the website [www.thegoodscentscompany.com](http://www.thegoodscentscompany.com), which states that Lyrall is not for flavor use.

For all of the above reasons, the applicant submits that Slangan does not anticipate any of the applicant's claims. Accordingly, withdrawal of the Section 102(b) rejection of claims 1-4 and 8-11 based on Slangan is requested.

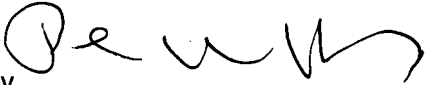
The Examiner is also requested to reconsider and withdraw the Section 103(a) rejection of claims 1-12 as obvious from the combination of Slangan et al. in view of "Common Fragrance and Flavor Materials" by Bauer et al. The applicants respectfully submit that there is no suggestion in the references, no matter how viewed, of the applicant's flavor compositions or products containing the same. The applicant's invention is directed to providing flavor compositions which taste of cinnamon without requiring cinnamic aldehyde. There is no disclosure or suggestion in either of the Examiner's references that the applicant's combination of cinnamic alcohol and eugenol would or could provide a useful cinnamon flavor composition which did not include or require cinnamic aldehyde. The applicant has provided such a composition and, with respect, it is submitted that the applicant's composition is both novel and unobvious. Accordingly, withdrawal of the Section 103(a) rejection is requested.

For the reasons stated, it is submitted that claims 1-12 define new and unobvious compositions or products based thereon. The same is true for new claim 13 which is even further removed from the art in view of its closed "consisting essentially of" language.

Favorable reconsideration with allowance is requested.

Respectfully submitted,

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Date: November 21, 2007

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## leerall

Right Click Picture For More Options. (Safari 1.2 (v125) Compatible).

4-(4-hydroxy-4-methylpentyl)  
cyclohex-3-ene-1-carbaldehyde

InChI=1/C13H22O2/c1-13(2,15)9-3-  
4-11-5-7-12(10-14)8-6-  
11/h5,10,12,15H,3-4,6-9H2,1-2H3

InChIKey : ORMHZBNNECIKOH-UHFFFAOYAY

cas number : 31906-04-4

ec number : 250-863-4

molar refractivity :  $63.23 \pm 0.3 \text{ cm}^3$

parachor :  $525.2 \pm 6.0 \text{ cm}^3$

index of refraction :  $1.527 \pm 0.02$

surface tension :  $42.6 \pm 3.0 \text{ dyne/cm}$

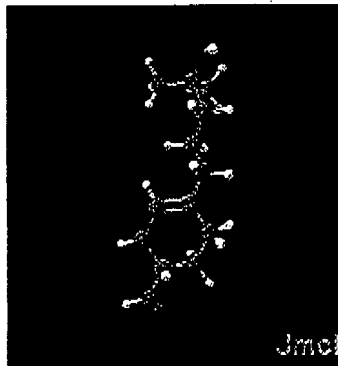
density :  $1.023 \pm 0.06 \text{ g/cm}^3$

polarizability :  $25.06 \pm 0.5 \text{ 10}^{-24} \text{ cm}^3$

xlogp : 2.50

molecular weight : 210.3125800

formula : C13 H22 O2



☐ Van der Waals surface

☐ Spin

fda reg :

h. number : 2912.30.0000

### organoleptics :

odor type : floral

odor strength : medium

odor description : at 100.00 %  
floral muguet cyclamen rhubarb woody

substantivity : 400 Hour(s)

### properties :

appearance : colorless to pale yellow clear viscous liquid

assay : 97.00 - 100.00 %

specific gravity : 0.98900 - 0.99700 @ 25.00 °C.

pounds per gallon - 8.229 to 8.296  
calc. :

refractive index : 1.48600 - 1.49300 @ 20.00 °C.

acid value : 5.00 max. KOH/g

logp : 2.53

shelf life : 24.00 month(s) or longer if stored properly.

storage : store in cool, dry place in tightly sealed containers, protected  
from heat and light. store under nitrogen.

### synonyms :

hydroxy-4-methyl  
4-(4-pentyl) cyclohex-3-ene  
carbaldehyde

hydroxy-4-methyl  
3-(4-pentyl) cyclohex-3-ene-  
1-carbaldehyde

3 and 4-  
(4- hydroxy-4-methyl  
pentyl)-3-cyclohexene-  
1-carboxaldehyde

hydroxy-4-methyl  
4-(4-pentyl)-3-cyclohexene-  
1-carboxaldehyde

hydroxyisohexyl 3-  
cyclohexene  
carboxaldehyde

hydroxymethyl pentyl  
4,4-cyclohexene  
carbaldehyde

kovanol

landolal

lyral

mugonal

### soluble in :

alcohol

water, very slightly

### insoluble in :

paraffin oil

water

### stability :

alcoholic fine fragrance,  
good

alcoholic lotion

antiperspirant, good

deodorant stick

detergent perborate

fabric softener, good

hard surface cleaner

liquid detergent

shampoo

soap, good

### (odor and/or flavor)

#### blends with :

acetaldehyde ethyl  
phenethyl acetal

acetate C-10

acetoin

alcohol C-10

alcohol C-11 undecylic

alcohol C-12

alcohol C-9

**safety :**

most important hazard (s) : Xi - Irritant

flash point ( Deg. F. ) : 200.00 °F. TCC ( 93.33 °C. )

IFRA critical Effect : Sensitization

limits in the finished product for - "leave on the skin contact" :  
1.5000 % Restriction.

limits in the finished product for - "wash off the skin contact" :  
1.5000 % Restriction.

limits in the finished product for - "no skin contact" :  
15.0000 % Restriction.

recommendation for usage levels up to :  
20.0000 % in the fragrance concentrate.

recommendation for usage levels up to :  
not for flavor use.

**safety links :**

msds : [msds](#)

toxnet : [31906-04-4](#)

epa-srs : [31906-04-4](#)

pubchem : [31906-04-4](#)

ifra : [ifra](#) - ( Found under : HMPCC )

**other :**

CofA

**references :**

aldehyde C-10

aldehyde C-11  
undecylenic

aldehyde C-12 lauroic

aldehyde C-12 myristic

aldehyde C-14

aldehyde C-16

aldehyde C-18

aldehyde C-9

algae absolute

allyl amyl glycolate

allyl cyclohexyl  
propionate

amber carbinol

ambroxan

iso amyl benzoate

alpha- amyl cinnamaldehyde

iso amyl salicylate

amyrin wood oil

angelica root oil

anise seed oil

para- anisyl acetate

para- anisyl alcohol

anise oil

bay leaf oil

beeswax absolute

benzoin resinoid

benzophenone

benzyl acetate

benzyl alcohol

benzyl cinnamate

benzyl isoeugenol

benzyl propionate

benzyl salicylate

bergamot oil

blood orange oil

bois de rose oil

iso bornyl acetate

laevo- bornyl acetate

iso butyl cinnamate

caraway seed oil

cardamom seed oil

camphor absolute

carrot seed oil

cassia bark oil

cedarwood oils

cinnamyl alcohol

citronellol

citronellyl acetate

# Perfume and Flavor Chemicals

(Aroma Chemicals)

## II

By

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CONSULTANT

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RUTGERS UNIVERSITY

THE STATE UNIVERSITY OF NEW JERSEY

UNIVERSITY EXTENSION DIVISION

NEWARK, N. J. (U.S.A.)

PUBLISHED BY THE AUTHOR 1969

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MONTCLAIR, N. J. (U.S.A.)

Steffen Arctander, publisher  
P.O. Box 114, Elizabeth, N.J. 07207, U.S.A.

Library of  
Congress Catalog Card Number 75-91398

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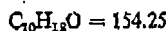
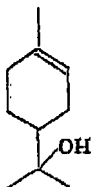
Printed in Denmark by Det Høfensbergiske Etablissement,  
Copenhagen V., Denmark



## 2877: alpha-TERPINEOL

*1-Methyl-4-iso-propyl-1-cyclohexen-8-ol*  
*para-Menth-1-en-8-ol*

The commercial perfume grade material is mainly *alpha*-Terpineol with 3 to 5% *beta*-Terpineol and traces of other alcohols.



Colorless, slightly viscous liquid.

B.P. 219° C. Solidifies in the cold, melts at 35° C. (with considerable variations according to purity). Sp.Gr. 0.94.

Very slightly soluble in water, soluble in alcohol, Propylene glycol, Mineral oil and perfume and flavor materials.

The odor of high-grade *alpha*-Terpineol being delicately floral and sweet of Lilac type, it is very strongly influenced in presence of minute amounts of impurities of different odor type. The most common undesirable odors are the more volatile terpenes and terpene alcohols (see *1*-Terpinenol) and the less volatile phenols, often occurring in traces in "European type" Terpineol. *beta*-Terpineol is the main impurity among the more volatile ones.

A "piney" odor from the more volatile impurities and a dry-medicinal odor from the less volatile ones are the most harmful olfactory notes to destroy the Lilac florality of perfumery grade Terpineol. It is common practise of the large users of perfumery grade *alpha*-Terpineol to purchase commercial grade Terpineol and purify their own perfumery grade from that.

Terpineol for flavor purpose should be of equally high olfactory (organoleptic) purity, except perhaps for Lime flavors, where certain qualities, unacceptable for a Lilac perfume, will be suitable for Lime flavor. But generally, a very high grade Terpineol is necessary.

*alpha*-Terpineol (or commercial Terpinol)

is one of the most commonly used of perfume chemicals. Its very low cost, excellent availability, general stability in air, as solvents and other chemicals, etc. and its versatility in compositions make it an everyday, all-purpose material in most perfume laboratories and compounding factories, though exact figures are not available. Annual world consumption of Terpineol for perfumery purposes can be estimated at less than 10,000 metric tons. However, only a relatively small portion of this volume is truly highgrade *alpha*-Terpineol, the balance being a commercial "high-*alpha*"-grade material, which in the U.S.A. is priced very near the cost of Diethylphthalate.

As a basic component in Lilac perfume and a major component of many varieties of Pine fragrances (lower grades of Terpineol will usually suffice) as common ingredient in Fougères, Appleblossoms, fragrances for household products, soaps, detergents, etc. it has very little competition.

Perfumers all over the world have for decades discussed the quality of Terpineol, and expressed their preference for one or the other type. It is understandable that the American perfumer is satisfied with the Terpineol most common to that continent, the Terpineol isolated from Pine oil, but it is not always easy to appreciate the terms "Lilac" for the *alpha*-Terpineol and "Hyacinth" for the *beta*-Terpineol.

The "European type" Terpinol, on the other hand, is rarely contaminated with *beta*-Terpineol (almost absent when Terpineol is made by acid hydrolysis of *alpha*-Pinene). The most annoying impurity in this type Terpineol is *iso*-Borneol, while traces of phenolic components may impair the terminal notes of that Terpineol.

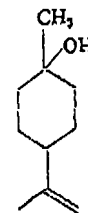
It has come so far, that customers have requested *beta*-Terpineol for perfumery purposes, often and unfortunately based upon analytical information dating back prior to the use of Gas-Liquid-Chromatography as a means of analysis in combination with Infra-red spectroscopy, Mass-spectrometry and perhaps Nuclear Magnetic Resonance readings.

It seems beyond doubt that *alpha*-Terpineol

is the most popular is known of no attempt is made at producing a *beta*-Terpineol in the what is commonly accepted. The acetate of *beta*-Terpineol is classified as "more acceptable" from *alpha*-Terpineol, and is welcome by the manufacturer himself left with a surplus after vacuum distillation. *alpha*-Terpineol is used in various flavor compounds, Lemon, Licorice, Ginger, Anise, Peach, for Frankfurt sauerkraut flavors. The concentration is as low as 5 to 40 ppm in Citrus and G.R.A.S. F.R.M. Prod.:

1) by isolation from

*Δ*-8,9-*para*-Menthane  
*1*-Methyl-4-*iso*-propyl-1-cyclohexen-8-ol  
The *trans-beta*-Terpinol form of *beta*-Terpineol



Colorless, slightly viscous liquid. Sp.Gr. 0.93. Solid again at 33° C. B.P.

Very slightly soluble in water and oils.

Pungent-woody odor of moderate tenacity.

The common odor is a bit difficult to first encounter with. See monograph.

commonly used of all per-  
very low cost, excellent  
stability in air, soap,  
chemicals, etc. and its  
ions make it an every-  
material in most perfume  
apounding factories. Al-  
are not available, the  
option of Terpineol for  
can be estimated at not  
ons. However, only a  
of this volume is  
the Terpineol, the bulk  
"high-alpha"-grade mate-  
S.A. is priced very near  
thalate.  
oient in Lilac perfumes  
ent of many varieties of  
ter grades of Terpineol  
as common ingredient of  
osoms, fragrances for  
soaps, detergents, etc. it  
ation.

if the world have for de-  
quality of Terpineol, and  
ference for one or the  
understandable that the  
is satisfied with the Ter-  
in to that continent, the  
com Pine oil, but it is not  
reciate the terms "Lilac"  
ool and "Hyacinth" for

"Terpineol, on the  
y, contaminated with *beta*-  
absent when Terpineol is  
hydrolysis of *alpha*-Pinene).  
is impurity in this type  
terneol, while traces of  
is may impair the terminal  
neol.

far, that customers have  
pineol for perfumery pur-  
unfortunately based upon  
tion dating back prior to  
aid-Chromatography as a  
in combination with Infra-  
Mass-spectrometry and per-  
cetic Resonance readings.  
doubt that *alpha*-Terpineol

is the most popular isomer, and the author  
knows of no attempt to promote a synthesis  
aimed at producing a higher proportion of  
*beta*-Terpineol in the reaction mixture than  
what is commonly accepted today.

The acetate of *beta*-Terpineol has been  
classified as "more interesting" than the  
acetate from *alpha*-Terpineol. This could only  
be welcome by the manufacturer who finds  
himself left with a surplus of the *beta*-isomer  
after vacuum distillation of the total Terpineol.

*alpha*-Terpineol is used in discrete amounts  
in various flavor compositions, such as berry  
flavors, Lemon, Lime, Nutmeg, Orange,  
Ginger, Anise, Peach, etc. and in spice flavors  
for Frankfurter sausages, or in "Lilac"  
flavors. The concentration used is normally  
as low as 5 to 40 ppm in the finished product,  
highest in Citrus and Spice compositions.

G.R.A.S. F.E.M.A. No.3045.

Prod.:

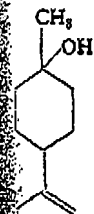
1) by isolation from American Pine oil.

## 2878: *Beta* TERPINEOL

4-8,9-*para*-Menthenol-1.

1-Methyl-4-*iso*-propenyl cyclohexan-1-ol.

The *trans-beta*-Terpineol is the most common  
form of *beta*-Terpineol.



$C_{10}H_{18}O = 154.25$

Colorless, slightly viscous liquid.

Sp.Gr. 0.93. Solidifies in the cold, melts  
again at 33° C. B.P. 210° C.

Very slightly soluble in water, soluble in al-  
cohol and oils.

Pungent-woody-earthy, but not piney, odor  
of moderate tenacity.

The common odor description "Hyacinth"  
is a bit difficult to appreciate, particularly at  
first encounter with the title material.

See monograph "*alpha*-Terpineol" (preced-

2) from *alpha*-Pinene via Terpin hydrate or  
directly to Terpineol by hydration.  
3) from Pentane tricarboxylic acid by cycliza-  
tion, followed by esterification, via the  
Hydroxyester to the unsaturated ester, and  
by Grignard reagent to Terpineol.

4) from Isoprene and Methyl vinyl ketone  
with Methyl magnesium iodide.

Methods No. 3) and 4) are "purely syn-  
thetic" using acyclic starting materials. They  
are of no commercial interest as long as raw  
materials for methods 1) and 2) are available  
in volume.

34-806; 67-517; 88-55; 104-537; 104-634;  
106-330; 140-127; 85-113; 156-328; 163-71;  
163-238; 163-377;

Gjilden Co., data sheet May 1961.

Hercules Powder Co., data sheets.

Arizona Chemical Co. data sheets.

ing this) for comments on the odor of the  
two isomers.

*beta*-Terpineol is rarely used as such in  
perfume compositions, but it is a common  
companion to the *alpha*-isomer in commer-  
cial grade Terpineol. It has also been suggested  
for the manufacture of Terpinyl acetate as it  
is preferred by some perfumers for that pur-  
pose.

It seems beyond doubt, that the *alpha*-  
Terpineol is the only suitable Terpineol for  
better grades of Lilac fragrances and other,  
similar florals, while the *beta*-Terpineol is no  
serious drawback in Pine fragrances, Fougères  
and ordinary fragrances for soaps, detergents  
and other household products.

The author has no report of this material  
being used in flavor compositions.

Prod.:

1) from fractions of Steam distilled Pine oil.  
2) from "heads" or foreruns of the rectifica-  
tion of Terpineol made via *alpha*-Pinene  
and Terpin hydrate.

3) by reduction of Limonene epoxide.

67-517; 88-82; 106-330; 156-328; 163-71;  
163-238; 163-377;